

Using Foam to Separate Ions: Superchaotropic Effects as a New Route toward Sustainable Chemistry

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The selective separation of ions in aqueous systems is a longstanding challenge in chemistry, environmental remediation, and resource recovery. We present a novel, solvent-free strategy that leverages *superchaotropic effects* to extract anionic metal species using foam flotation.

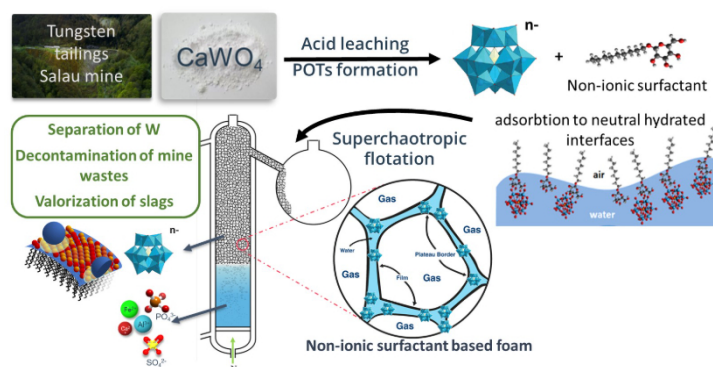
Unlike classical ion flotation, which requires charged surfactants, superchaotropic nano-ions, such as some polyoxomolybdates and polyoxotungstates, adsorb efficiently onto non-ionic surfactant films. ^[1] This unexpected affinity allows their selective enrichment in foam under mild conditions, providing a sustainable platform with minimal organic content. ^[2]

We illustrate the feasibility of superchaotropic flotation with two systems. Molybdenum forms iso-polyoxomolybdates at low pH whose superchaotropic character drives foam-mediated separation. ^[2] Tungsten forms heteropolyoxotungstates with phosphate, which are also superchaotropic; importantly, these species can be generated from real mining residues, demonstrating the potential for environmental remediation and the valorization of industrial waste. ^[3]

These results demonstrate how the fundamental concept of superchaotropicity, introduced in 2015 ^[1, 4], can now be harnessed for applied separation chemistry, bridging interfacial science with sustainable, solvent-free technologies.

Keywords:

foam, flotation, ion, separation, solvent-free



Example of superchaotropic flotation for the recovery and valorization of tungsten mine tailings

References:

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