

Colloidal and Interfacial Properties of Silicon Nanocrystals and Their Impact on Lithium Ion Battery Performance

Fatima Hassouna ^a, Gabriela Soukupová ^a, Pavel Galář ², Filip Matějka ^{a,b}, Zuzana Vlčková Živcová ^c, Otakar Frank ^c, Jiří Červenka ^b

^(a) Faculty of Chemical Engineering, University of Chemistry and Technology, Prague, 166 28 Prague 6, Czech Republic

^(b) FZU - Institute of Physics of the Czech Academy of Sciences, Cukrovarnická 10/112, Prague, 162 00 Prague 6, Czech Republic

^(c) J. Heyrovsky Institute of Physical Chemistry, Czech Academy of Sciences, Dolejškova 2155-3, Prague 18223 8, Czech Republic

fatima.hassouna@vscht.cz

Silicon nanocrystals (SiNCs) are promising candidates for next-generation energy and hybrid materials, yet their practical integration is often limited by interfacial instability, aggregation, and sensitivity to aqueous environments. In this work, we investigate SiNCs of 6, 20, 55, and 100 nm to understand how surface chemistry, particle size, and solid-state properties jointly influence their colloidal behaviour and functional performance. Using hydrogen-terminated SiNCs as a starting point [1], we show that functionalization with poly(acrylic acid) (PAA) and phytic acid (PhA) produces covalent and ionic Si–O–C and Si–O–P linkages that substantially enhance dispersion stability. Spectroscopic analysis (FTIR, XPS), dynamic light scattering, and zeta-potential measurements demonstrate that controlled surface chemistry stabilizes surface charge and suppresses aggregation across a broad pH range, highlighting the central role of interfacial speciation in colloidal control.

Extending this strategy across all investigated sizes enables a direct comparison of how curvature, accessible reactive sites, and initial oxidation state affect functionalization efficiency and colloidal stability. Incorporation of size-defined SiNCs into hydrogel-derived polypyrrole (PPy) networks further reveals the interplay between colloidal stabilization and electrochemical behaviour [2]. Smaller SiNCs show the most favourable electrochemical response, achieving capacity retention up to 85% after 100 cycles, whereas physically blended electrodes display weaker cohesion and accelerated degradation. Electrochemical analyses (GCD, CV, EIS) confirm that optimised interfacial environments enhance charge-transfer stability and mitigate irreversible degradation pathways.

Overall, this work provides a unified framework demonstrating how surface functionalization, size effects, and solid-state properties together determine the colloidal and electrochemical performance of Si nanomaterials. These insights support the development of stable, high-performing Si-based colloids relevant to colloid and interface science, sustainable processing, and next-generation lithium-ion energy technologies.

References:

- [1] Pavel Galář; Gabriela Čandová; Filip Matějka; Fatima Hassouna; Abdelghani Laachachi; Petr Sajdl; Kateřina Kusová. Green and scalable surface functionalization of silicon quantum dots using water-soluble organic acids for sustainable hybrid materials. *RSC Advances* 2026, 16, 2923–2936.
- [2] Gabriela Soukupová; Filip Matějka; Zuzana Vlčková Živcová; Abdelghani Laachachi; Pavel Galář; Miloslav Lhotka; Otakar Frank; Jiří Červenka; Fatima Hassouna. Tailored silicon nanostructures in hydrogel-derived conductive binders: Role of size, structure, and surface chemistry in enhancing Li-ion battery performance. *Journal of Power Sources* **2026**, 661, 238620.

Acknowledgements:

The infrastructure used was made available through project No. CZ.02.01.01/00/22_008/0004617 “Energy conversion and storage (ECO&Stor)”, funded by the European Union and the state budget of the Czech Republic within the framework of the Jan Amos Komenský Operational Program.