

Confining Geometry Drives Dynamics in Lipidic Mesophases

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In biological and advanced nanomaterial environments, water and other non-aqueous solvents often exist in confined settings, where dynamics are strongly coupled to surrounding interfaces [1]. While the size of confinement plays a major role, the effect of confining geometry on solvent dynamics remains an open question [2,3]. Lipidic mesophases (LMPs) serve as a versatile platform to study soft confinement effects by offering a range of self-assembled phase structures with nanometer-sized solvent-domains [4]. By tuning composition and temperature, LMPs transition from structures with planar interfaces to curved interfaces with interconnected solvent channels, and further to highly curved, isotropic phases, such as the lamellar (L_α), cubic $Ia3d$, and reverse micellar (L_2) phases, respectively. Combined X-Ray Scattering and Calorimetry reveal that transitions between these phases occur with modest enthalpic input but trigger discontinuous shifts in lipid- and solvent-dynamics, as probed by complementary spectroscopic techniques: In both water- and glycerol-based LMPs, conductivity increases when systems transition into more fluid phases and solvent-connectivity is increased. H-bonding strength increases as interfaces curve towards planar membranes, revealing dynamically different interfacial and core water populations, while lipid tail packing tightens. Together, these findings establish a clear geometric principle: interfacial curvature and solvent connectivity jointly control the packing of lipids, hydrogen-bonding, solvent dynamics, and charge transport in lipidic mesophase, providing a conceptual framework for the rational design of lipid-based systems.

Keywords:

lipidic mesophase, phase structure, dynamics, spectroscopy

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